UNIVERSITAS NEGERI YOGYAKARTA
FACULTY OF MATHEMATICS AND NATURAL SCIENCES
DEPARTMENT OF CHEMISTRY EDUCATION
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Bachelor of Education in Chemistry **MODULE HANDBOOK**

Module name:	**Materials Development on Chemistry**
Module level, if applicable:	Undergraduate
Code:	MPK 6215
Sub-heading, if applicable:	-
Classes, if applicable:	1
Semester:	Even
Module coordinator:	Dr. Das Salirawati
Lecturer(s):	**Marfuatun, S.Pd.Si.,M.Si.**
Language:	Bahasa Indonesia
Classification within the curriculum:	Elective Course
Teaching format / class hours per week during the semester:	100 minutes lectures, 120 minutes structured activities, and 120 minutes individual per week.
Workload:	Total workload is 90.67 hours per semester which consists of 100 minutes lectures, 120 minutes structured activities, and 120 minutes individual study per week for 16 weeks.
Credit points:	2SKS (3.28ECTS)
Prerequisites course(s):	-
Course Outcomes	After taking this course the students have ability to: CO1. implement procedures in developing teaching materials CO2. explain the concept of teaching materials in chemistry learning CO3. understand the procedures for developing teaching materials CO4. compile teaching materials according to the concept of developing correct teaching materials
Content:	This course is an elective course with the aim of students being able to develop chemical teaching materials with current and up-to-date sources. This course includes material on the preparation of design, development, management and evaluation of teaching materials. The lecture implementation uses an active learning model with lecture, question and answer, discussion, and project learning methods.
Study / exam achievements:	Attitude assessment is carried out at each meeting by observation and / or self-assessment techniques using the assumption that basically every student has a good attitude. The student is given a value of very good or not good attitude if they show it significantly compared to other students in general. The result of attitude assessment is not a component of the final grades, but as one of the requirements to pass the course. Students will pass from this course if at least have a good attitude. The final mark will be weight as follow:

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<tr><th>No</th><th>CO</th><th>Assessment Object</th><th>Assessment Technique</th><th>Weight</th></tr>
<tr><td>1</td><td>CO1
CO2
CO3
CO4.</td><td>Activities
Assignments
Mid-term exam
Final Exam</td><td>Presentation / written test</td><td>10%
40%
25%
25%</td></tr>
<tr><td colspan="4">Total</td><td>100%</td></tr>
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<tr><td>Forms of media:</td><td>Board, LCD Projector, Laptop/Computer</td></tr>
<tr><td>Literature:</td><td>

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PLO and CO mapping

	PLO					
	Attitude		Knowledge	Specific Skill	General Skill	
	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6
CO1				√		
CO2			√			
CO3			√			
CO4					√	